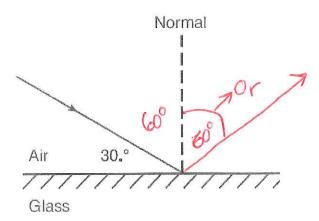
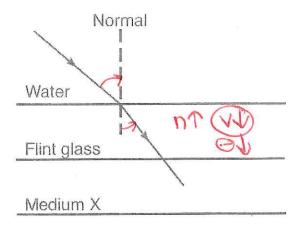
	My mistake-Toundingenor		
408	What is the speed of a wave with a	416	Determine the energy of photon with a
400	wavelength of 0.3 m and a frequency of 12	410	frequency of 5.09 x 10 ¹⁴ Hz.
	Hz?		a) 1.6 x 10 ⁻¹⁹ J c) 3.4 x 10 ⁻¹⁹ J
Da	Western St.		b) $3.0 \times 10^8 \text{ J}$ d) $5.9 \times 10^{-7} \text{J}$
	a) 27 m/s c) 4 m/s		
	b) 60 m/s d) 3.0 x10 ⁸ m/s	2020	E=hf=(6.63x1034)s(5,09x1047)=3,37x107
	V=FX=(DH=21/3m)=3,6ms	417	Determine the energy in electronvolts of photon
409	What is the speed of light with a frequency of		with a wavelength of 5 x 10 ⁻⁷ m
	5.09 x 10 ¹⁴ Hz when traveling through water?		a) 4.0 x 10 ⁻¹⁹ eV c) 2.5 eV
	a) $5.9 \times 10^{14} \text{ m/s}$ c) $2.25 \times 10^8 \text{ m/s}$		b) $6 \times 10^{-14} \text{eV}$ d) $3.8 \times 10^{33} \text{eV}$
	b) $3.0 \times 10^8 \text{ m/s}$ d) $4.4 \times 10^8 \text{ m/s}$		F=hs=3.978×10-19) -1.6×10-18/64=2.486eV
	b) 3.0 x 10 ⁸ m/s d) 4.4 x 10 ⁸ m/s	418	Light with a frequency of 5.09 x 10 ¹⁴ Hz in air is
410	Which color of light has a wavelength of 5.0 x	3	incident on a boundary with water at an angle of
	10 ⁻⁷ m in air?		30° What is the angle of refraction?
	a) Blue c) Orange		a) 15.4° c) 0° $ SinC = SinC $ b) 45° d) 22° $ SinC = SinC $
			b) 450 Sm30=1.335m02
	b) Green d) Violet	440	b) 45° d) 22° Eight with a frequency of 5.09×10^{14} Hz in air is 6.22°
444	7 5x10 m 6x10 HZ	419	Light with a frequency of 5.09 x 10 Hz in air is
411	Which characteristic is the same for every		incident on a boundary with another substance.
	color of light in a vacuum?		The angle of incidence is 45° and the angle of
	a) Frequency c) Speed		refraction is 30°. What is the index of refraction
	b) wavelength d) Period		
			for the substance? $N_1 Sine_1 = N_2 Sine_2$
			15/143-1125/1130
			a) 1.4 c) 1.66 h) 0.7
412	In which way does blue light change as it	420	b) 0.7 d) 1.33
712	travels from crown glass into diamond?	720	
	The control of the co	1	and the first state of the stat
	O COLONIA DE	1.52	What is the minimum amount of energy
	b) Its frequency increases	242	required to ionize an electron in the ground
			state of a mercury atom?
	d) Its speed increases		a) 13.6 eV c) 4.64 eV
	SOVV		b) 10.38 ev d) 20 eV
			n=a=10.38eV
413	The slope of a graph of photon energy vs.	421	Determine the energy emitted by an electron as
	frequency equals		it falls from n=3 to n=2 in a hydrogen atom.
	a) Photon speed		a) 12.09 eV c) 15.11 eV
	b) Wavelength		(b) 1.89 eV (d) 1.51 eV
	C) Planck's constant		n=3=1,5kVZ 1.89eV
	d) Speed of light squared		1-3-10R1 & 1089eV
	a) Speed of light squared		12=340eV)
414	How much anarquis produced from the	422	A singer breaking a glass by hitting a specific
414	How much energy is produced from the complete conversion of an electron into	444	A singer breaking a glass by hitting a specific
			frequency is known as the phenomena of
Sec. 1	energy?		a) Diffraction c) Interference
	a) 1.6×10^{-19} J c) 3.2×10^{-19} J		b) Doppler effect d) Resonance
	b) 2/1×10 ²² d) 8.2 x 10 ⁻¹⁴ J		
	F=mc=9 (1x103/k) 3x108/m/k)		
	L THE TAR	423	Maximum constructive interference occurs
	The same of the sa		when two waves
415	How much energy is produced when 2		
	universal mass units are converted into		a) Are in phase
	energy?		b) Out of phase by 90°
	(a) $1.86 \times 10^3 \text{ MeV}$ c) $3.0 \times 10^8 \text{ MeV}$		c) Out of phase by 180°
	b) 2.8 x 10 ³ MeV d) 1.6 x 10 ⁻¹⁹ MeV		d) Out of phase by 45 ⁰
		N	
	Qu x (3/x103MeV)=186x103N	le v	

Draw the reflected ray on the diagram below. 424 Label the angle of reflection including value of angle

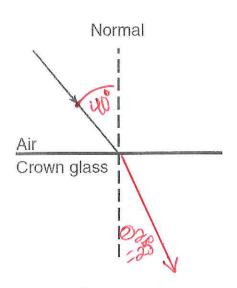
What happens to the speed of the light wave as it moves from water to flint glass?





Velocity obcreases

426.



For the light ray to the left

a) Use a protractor to determine the angle of incidence [1]

b) Determine the angle of refraction (show all work including equation, and substitution with units) [2]

N, Sine, = N, Sine, 1 Sm46 = 1,52 Sine,

- c) Draw in the angle of refraction on the diagram at left [1]
- A photon with a frequency of 6.0 x10¹⁴ hertz is absorbed by an excited hydrogen atom. 427 This causes the electron to be ejected from the atom, forming an ion.
 - a) Calculate the energy of this photon in joules. [Show all work, including the equation and substitution with units.] [2]

425

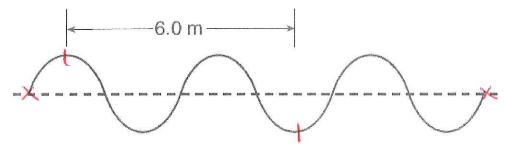
E=h-P (6.63x10-34/s)(6x104/tz) = 3.98x10-19)

- b) Determine the energy of this photon in electronvolts. [1]

 3.8×10-43× 10-43 = 2.486eV

 c) What was the initial energy level of the electron before being ejected from the atom by the photon?

The initial level must have been level 3 n=1.5/eV (n = 2 would prove been too low)



d) For the wave pictured above, determine the wavelength of a single cycle [1]

2 = meters = 6m = attm

e) If the entire wave train above took 12 seconds to pass, what is the frequency of the wave? (show work including equation, substitution and units) [2]

P= Cycles = 3 cycles = 25Hz

Determine the speed of the wave (show all work including equation, substitution and units)? [2]

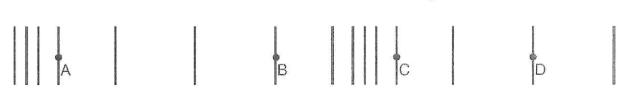
1= (25Hz)=: - |m/s

g) What type of wave motion is pictured in the diagram above? [1]

transverse

A sound wave moving through STP is pictured below.

Wave movement



- What type of wave motion is represented in the picture above?

 List two points (by letter) that are one wavelength apart.

A & (

c) What is the speed of this wave at STP?

Soundwaves more 331ms at STP

d) What is the speed of this wave in a vacuum?

It can't move in a vacuum